



Technology Platform for Sampling Water with Electrolyte-Gated Organic Transistors Sensitised with Langmuir-Deposited Calixarene Surface Layers

Delia Puzzovio^{1,*}, Abdullah Al Naim¹, Lee Hague¹, Mary Deasy²,
James Ward², Tim Richardson¹, and Martin Grell¹

¹Physics and Astronomy, University of Sheffield, Hicks Building, Hounsfield Rd, Sheffield S3 7RH, UK

²Department of Science, Institute of Technology Tallaght (ITT Dublin), Dublin 24, Ireland

We demonstrate a technology platform that enables the development of new, surface-sensitised organic transistor sensors. We show that an organic semiconductor can still be gated by an electric double layer within the electrochemical window of water after the deposition of up to four Langmuir-Schäfer calixarene layers onto its surface. Since many calixarenes are known to selectively bind waterborne cations, this facilitates sensitising a conventional organic semiconductor with a physically deposited layer for specific cation recognition. When at least two Langmuir-Schäfer layers are deposited, these also block the electrochemical doping of the organic semiconductor, which otherwise competes with the field effect in water-gated organic transistors. Carrier mobility is reduced by the application of calixarene layers, but transistor current measurement remains accessible by simple methods. We find that for the present purpose, Langmuir-Schäfer-printed surface layers perform better than those deposited by Langmuir-Blodgett deposition.


Keywords: Polythiophene, Organic Transistor, Water, Electrolyte, Electric Double Layer, Langmuir-Blodgett, Langmuir-Schäfer, Calixarene, Surface, Sensitiser, Sensor.

Technology Platform for Sampling Water with Electrolyte-Gated Organic Transistors Sensitised with Langmuir-Deposited Calixarene Surface Layers

Delia Puzzovio, Abdullah Al Naim, Lee Hague, Mary Deasy, James Ward, Tim Richardson, and Martin Grell
J. Surf. Interfac. Mater. 1, 49-54 (2012)

<http://www.aspbs.com/jsim.htm#v1n1>

Second Paper:



Home Browse ▾ About ▾ Authors ▾ Librarians ▾ Features ▾ Purchase

Applied Physics Letters / Volume 101 / Issue 14 / SURFACES AND INTERFACES

Appl. Phys. Lett. **101**, 141603 (2012); <http://dx.doi.org/10.1063/1.4757131> (4 pages)

Electron transporting water-gated thin film transistors

Abdullah Al Naim and Martin Grell
Department of Physics and Astronomy, University of Sheffield, Hicks Building, Hounsfield Road, Sheffield, South Yorkshire S3 7RH, United Kingdom
[View Map](#)

(Received 5 April 2012; accepted 19 September 2012; published online 2 October 2012)

http://apl.aip.org/resource/1/applab/v101/i14/p141603_s1?isAuthorized=no

Third Paper:



Journal of Applied Physics / Volume 112 / Issue 11 / ARTICLES / Device Physics

J. Appl. Phys. 112, 114502 (2012); <http://dx.doi.org/10.1063/1.4768203> (5 pages)

 CrossMark
← click for updates

Organic solvents as gate media for thin-film transistors

Abdullah F. Al Naim and Martin Grell
Physics and Astronomy, University of Sheffield, Hounsfield Rd, Sheffield S3 7RH, United Kingdom
[View Map](#)

(Received 2 August 2012; accepted 31 October 2012; published online 4 December 2012)

http://jap.aip.org/resource/1/japiau/v112/i11/p114502_s1?isAuthorized=no



Fourth Paper:



Organic Electronics

Volume 14, Issue 4, April 2013, Pages 1057–1063



Water-gated organic nanowire transistors

The remaining authors report with great sadness the passing of our co-author Tim Richardson shortly before publication. We wish to dedicate this work to the memory of a much missed colleague and friend.

Abdullah Al Naim^a,  , Adam Hobson^a, Richard T. Grant^a, Antonis Dragoneas^a, Mark Hampton^b, Chris Dunscombe^b, Tim Richardson^a, J. Emyr Macdonald^b, Martin Grell^a

^a Department of Physics and Astronomy, University of Sheffield, Hicks Building, Hounsfield Road, Sheffield, South Yorkshire S3 7RH, United Kingdom

^b School of Physics and Astronomy, Cardiff University, The Parade, Cardiff CF24 3AA, United Kingdom

<http://dx.doi.org/10.1016/j.orgel.2013.01.024>, How to Cite or Link Using DOI

<http://www.sciencedirect.com/science/article/pii/S1566119913000426>

First Poster (presented in SID on 17-18th Sep. 2012, Imperial Collage, London, UK) and (presented in NPL on 12th Dec. 2012, Surrey, London, UK)



Progress in electrolyte-gated thin film transistors

Abdullah F Al Naim, Martin Grell

Department of Physics and Astronomy, University of Sheffield, Hicks Building, Hounsfield Road, Sheffield, South Yorkshire, United Kingdom, S3 7RH

SID ORGANIC ELECTRONICS
UK 2012
Imperial College London
17-18 September 2012

1. Introduction

In recent years, it has been shown that thin film transistors (TFTs) can be gated via an electrolyte due to the high capacitance electric double layer (EDL) that forms at the electrolyte/semiconductor interface under applied gate bias. In particular the observation of Kerpoat *et al.*[1] that an organic TFT can be gated by pure-water has generated a great interest in using such devices in the sensing of aqueous-analytes.

Here, we demonstrate an electron-transporting water-gated TFT, using thermally converted precursor-route Zinc Oxide (ZnO), as seen in fig.1. Also, we show that some organic solvents, can act as EDL gate media for TFTs, in a similar way as solid electrolytes, ionic liquids, and water itself. We demonstrate this for both p- and n-type TFTs. Finally, we demonstrate electrolyte-gated organic NW field-effect TFTs using NWs grown from both p-type, and n-type, Organic Semi-conductors OSCs such as nano-belts of BBL for n-type, and nano-wires of PBTTT and P3HT for p-type.

2. Materials and Method

Zinc acetate dihydrate ($Zn(OAc)_2$) in ethanol is converted into ZnO. It shows ZnO here is largely an intrinsic semiconductor, just like sputtered ZnO [2]. We have treated the ZnO surface of such samples to turn them hydrophobic. This was either by spin casting HMDS or one layer of stearic acid by the Langmuir-Schäfer (LS) technique [3]. For gating n-type nano-belts, we have selected poly(benzimidazobenzophenanthroline) (BBL) that possesses a rather deep lowest unoccupied molecular orbital (LUMO) of -4.0 eV [4]. As suggested from Nicolai H.T. *et al.* [5], the only way to avoid traps is to possess polymers that its LUMO levels are lower than 3.6 eV below vacuum level. Nanowire growth: 'CB-P3HT' NWs were grown in Chlorobenzene poly(3-hexylthiophene), mP3HT, and 'Anisole-P3HT' NWs were prepared via a variation on the whisker method [6]. (sourced from Ossila)

3. Overview Fabrication

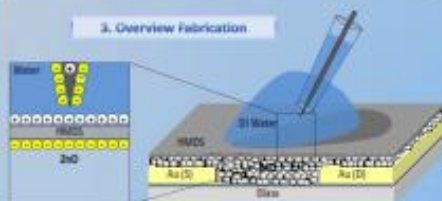


Fig. 1: Schematic illustration of the water gating setup, as described. Inset: Positive lines applied to gate needles leads to an electron accumulator layer at the ZnO surface. Also the anionic EDL in the water near the gate needles, and the cationic EDL in the water near the ZnO surface

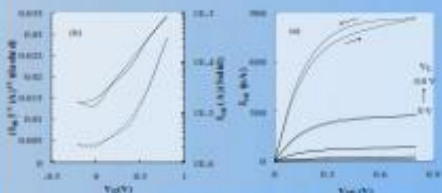


Fig. 2: Output (I_d)-and saturated transfer (I_{sat}) characteristics of water-gated HMDS-ZnO TFTs. In Fig. 3b, the source-drain voltage for the saturated transfer characteristics was 0.9 V.

4. Processing and Characterisation

For Isotherm and LS deposition, a NIMA Langmuir trough (Model 611) was used. To record TFT output- and transfer characteristics, we used two Keithley 2400 source/measure units. IV converter unit is used to measure the real-time characterisation of on-current, off-current and other vital parameters such as threshold and mobility, as described in [7, 8].



Fig. 3: Sinusoidal drive voltage V_G (amplitude 0.8V, $f = 1$ Hz), and resulting saturated drain current (I_{sat}) for ZnO TFTs gated with water, and several organic solvents. I_{sat} under water-gating is downscaled 10-fold to fit on the same scale.

	Water	Methanol	IPA	Acetone	Acetonitrile	Chloroform	Toluene	Cyclohexane
pKa	14	16.6	20.7	>32.5	-	-	-	-
Dipole [D]	1.85	1.7	1.66	2.88	3.92	1.84	0.36	0
H ₂ O sol. [wt%]	-	misc.	misc.	misc.	misc.	8	0.47	-8
ZnO	Y-	Y-	Y-	Y-	Y-	N	N	N
I_{sat} [%]	100	2.5	28.5	3.8	5.1	0	0	0
P3HT	Y+	Y+	Y+	Y+	Y+	X	X	N
I_{sat} [%]	100	12.7	8.5	5.8	11.8	X	X	0
PBTTT	Y+	Y+	Y+	Y+	Y+	X	X	N
I_{sat} [%]	100	13.2	6.6	5.3	14.3	X	X	0

Table 1: Summary of physicochemical properties of the solvents used here, and results for attempting to gate different semiconductors with different solvents. pKa for autoprotolysis at 25 °C from [9]; solvents that cannot undergo autoprotolysis have infinite pKa. Molecular dipole moments (in Debye) from [10]. Solubility in water in g of solvent per Liter of water; 'misc.' indicates miscibility in any ratio; Y-/Y+: Electron / hole field effect current was observed under applied gate voltage. (Y*) Small hole current. N: No current.

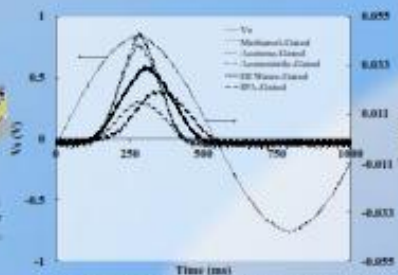


Fig. 4: Sinusoidal drive voltage V_G (amplitude 0.8V, $f = 1$ Hz), and resulting saturated drain current (I_{sat}) for PBTTT TFTs gated with water, and several organic solvents. I_{sat} under water-gating is downscaled 10-fold to fit on the same scale.

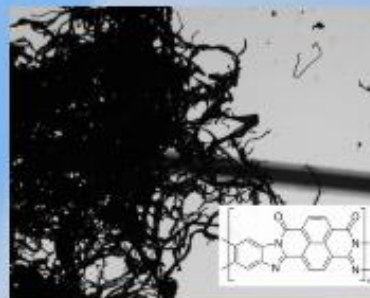


Fig. 5: Optical micrograph of BBL 'nanobelt' on TFT substrate, and twin-belted nanobelt from increases width/diameter of nanobelt. It has been deposited on gold contacts that thermally evaporated on SiO₂. Inset: the molecular structure of BBL that been used here.

5. Conclusion

- Water gated HMDS-ZnO TFTs display low threshold and high electron mobility. Moreover, Gate geometry is relevant for optimum performance of water-gated TFTs.
- We recommend the use of L-shaped electrodes that overlap the width of the TFT channel with leaving a water gap in the order of 0.1 mm.
- We find that organic solvents that are miscible with water (i.e., polar organic solvents) can act as EDL gate media for thin film transistors, in a similar way as previously reported for water. The most likely explanation for the ability to act as EDL gate medium is the inevitable presence of trace amounts of dissolved salts in such solvents.
- Acetonitrile in particular is an attractive alternative gate medium to water.
- Acetonitrile gating avoids -OH groups in the gate medium that are known to act as charge carrier traps [11], and offers a wider electrochemical window than water [12].
- We have demonstrated the first water-gated n-type and p-type organic nanowire field-effect devices.
- For the application of interfacial sensitizers, the polar (hydrophilic) ZnO surface may be an advantage over organic semiconductor surfaces. Electron transporting TFT transducers would be particularly suited for the detection of cationic waterborne species.

6. Acknowledgements

Abdullah F Al Naim acknowledges financial support from the Saudi Cultural Bureau in the UK, London and King Fahad University in Saudi Arabia, Alhassa.

References

- [1] Kerpoat, L; Hergaud, D; Buge, B; Pire, M; C. Pham, F; Cingol, M; Berggren, G; Horowitz, Adv Mater. 22, 2965 (2010).
- [2] E. M. C. Fortunato, F. M. C. Semadeni, A. C. M. S. G. Pimentel, A. M. T. Gomes, A. J. S. Marques, L. M. N. Pereira, F. F. Martins, Adv Mater 17, 290 (2005).
- [3] Abdullah Al Naim, Martin Grell, Electron-transporting water-gated thin film transistors, Applied Physics Letter, Accepted.
- [4] Alejandro L. Briseno, Felix Sanyal Elm, Amir Baheti, Younan Shi, and Samson A. Jenekhe, Journal of Materials Chemistry 21 (41), 15481 (2011).
- [5] H. T. Nicolai, M. Kulk, G. A. P. Wever, S. de Boer, C. Campbell, C. Rios, J. L. Brédas, and F. W. M. Steun, Nat Mater advance online publication (2012).
- [6] K. J. He, J. Mouton, P. Joubert, J. Polym. Sci., Pol. Phys. 31, 705 (1993).
- [7] L. Hagan, G. Pizzuto, A. Dragomir, M. Grell, Sci. Adv. Mater. 2(1), 5, 907.
- [8] G. Pizzuto, A. Al Naim, L. Hagan, M. Grell, I. Ward, T. Richardson, M. Grell, J. Appl. Phys. Mater. 2012, 1, 1, DOI 10.1063/1.3212120.
- [9] S. Rindfleisch, P. Joubert, P. R. Moore, T. Mousty, Paris & Appl. Chem. 39, 1995 (1987).
- [10] Reddy, L.A., Burger, W.B., and Sauer, T.K. Organic Solvents, Fourth Edition, John Wiley & Sons, New York, 2006.
- [11] H. B. James, A. Facchetti, M. R. Wasielewski, T. Marks, ACS 126, 12329 (2007); (b) W. F. Lee, J. H. Oh, S. L. Surin, W. C. Chen, F. Wudl, J. Phys. Chem. Adv. Funct. Mater. 21, 4273 (2012).
- [12] A. J. Bard and L. R. Faulkner, Electrochemical Methods: Fundamentals and Applications (2nd ed.), 2001 John Wiley & Sons Inc.

Second Poster: (presented by my colleague in 2012 MRS Fall Meeting & Exhibit on November 25 - 30, 2012, Boston, MA, USA).



Polythiophene nanowire thin-film devices



A. Dragoneas^{1*}, A. F. M. Al Naim¹, A. Hobson¹, T. Richardson¹, M. Grell¹, M. Hampton², J. E. Macdonald²,

¹ The Department of Physics and Astronomy, The University of Sheffield, S3 7RH, Sheffield, UK

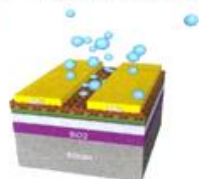
² School of Physics and Astronomy, Cardiff University, Queens Buildings, The Parade, Cardiff CF24 3AA, UK

*a.dragoneas@sheffield.ac.uk

Introduction

Polymers such as polythiophenes have gained notable scientific interest thanks to their important semiconducting and optoelectronic properties. The respective research work has yielded remarkable results and numerous applications over the last years^[1]. Prominent examples are the thin-film devices, such as chemiresistors, diodes and field-effect transistors. Their demonstrated applications span from the fields of light emission and photovoltaics to sensor devices. More recently, one-dimensional crystals of the same organic materials, which are usually referred to as 'organic nanowires', have broadened the spectrum of potential applications and spurred a new research interest^[2]. Differing only in terms of their morphology, organic nanowires can exhibit interesting practical advantages over conventional thin films of the same material, especially in the field of sensors. Here we present a study on devices mainly made of poly(hexylthiophene) (P3HT) nanowires, as well as a comparison of them to their respective conventional thin film counterparts. Solutions of both forms of these materials were produced and then spincast for the fabrication of devices of identical structure and geometry. The tested device architectures comprised chemiresistors, as well as thin-gate-dielectric and water-gated low-voltage transistors. The devices underwent electrical characterisation, atomic force microscope imaging and, in the case of sensing, exposure to controlled humidity conditions. The measurement results show the resemblance in the electrical properties of the two different forms of the materials, whereas they highlight notable behavioural variations, such as difference in moisture sensitivity, which can be attributed to the particularities of the nanowire morphology. Moreover, based on these findings, we introduce new prospects for future sensing applications of the organic nanowire devices.

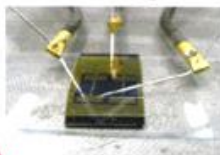
Nanowire OFETs for vapour sensing applications



Low-voltage organic field-effect transistors (OFETs) with polythiophene nanowires as active material have been built.

A thermally evaporated aluminium layer which is partially anodised to the formation of a thin oxide has served as the gate electrode and dielectric^[3]. The oxide was treated with an octadecylchlorosilane (OTS) self-assembled monolayer (SAM).

Anisole solutions of poly(hexylthiophene) (P3HT) and poly(3,3'-didodecylquaterthiophene) (PQ4T-12) NW were spincast on the treated oxide and gold contacts were thermally evaporated on top. This architecture can be deployed on a flexible substrate.



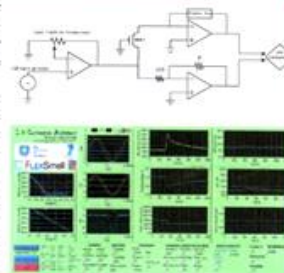
Although the nanowire devices exhibit drain current magnitudes which are usually smaller but comparable to the ones measured for conventional thin film control devices, their larger surface area is promising for sensing applications where the interaction with an airborne analyte takes place on the surface of the sensor^[4].

Real-time electrical characterisation of thin-film devices^[5]

In the pursuit of a full exploitation of the sensing capabilities of the polythiophene thin-film OFETs and chemiresistors, a new electrical characterisation scheme has been developed alongside these devices.

This system is based on bespoke readout electronics and a LabVIEW application. It is capable of calculating all the important transistor parameters in real-time making the detection of quick sensor responses possible.

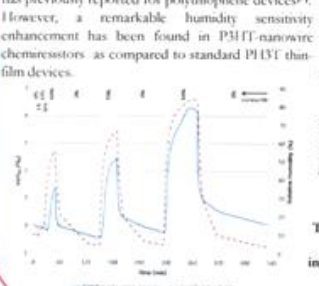
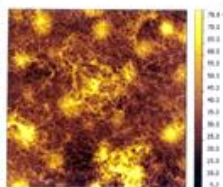
In the particular case of OFETs, the detection of changes in multiple parameters, such as their threshold voltage and the mobility of carriers, can provide more information on the analyte under investigation and effectively enhance the selectivity of the sensor.



Humidity sensitivity of P3HT-nanowire chemiresistors^[6]

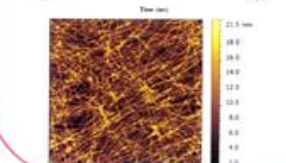
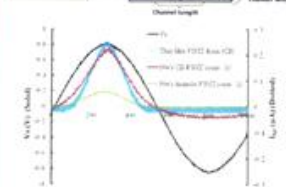
Resistive sensors (chemiresistors) were made for the comparison of standard P3HT film to P3HT nanowire mesh devices. Anisole solutions of both P3HT morphologies were prepared and spincast onto octadecylchlorosilane (OTS)-treated substrates. Gold contacts were thermally evaporated through shadow masks on top.

The devices were tested in a gas rig in a synthetic air environment with controlled humidity. The conductance of the chemiresistors of both morphologies was comparable under dry conditions. In all cases, the conductance was found to increase with increasing relative humidity (RH) levels, as it has previously reported for polythiophene devices^[7]. However, a remarkable humidity sensitivity enhancement has been found in P3HT-nanowire chemiresistors as compared to standard P3HT thin-film devices.



The nanowire-based chemiresistors exhibit a 75-fold conductance increase under 90% RH as compared to the dry state

Water-gated nanowire OFETs



P3HT nanowire (NW) solutions were prepared following two different routes, in which chlorobenzene (CB) and anisole were used as solvents, respectively. A CB solution of standard P3HT was also prepared.

The three solutions were spincast on silicon substrates with patterned gold contacts. A droplet of deionised (DI) water was used as the gate medium and a tungsten needle as the gate contact.

The NW devices performed as transistors with a good on-off current ratio and a saturated current which was found to be roughly an order-of-magnitude smaller than the conventional thin film device.

The CB-NW morphology yielded higher drain current as compared to the anisole-NW ones, at the expense of a higher static current.

There is ongoing work on the exploitation of these devices as sensors for waterborne analytes.

References

1. M. J. Joung *et al.*, *Synthetic Metals*, **149**, 75-77 (2005)
2. A. L. Briseno *et al.*, *Nature Today*, **41**, 4, pp.38-47 (2008)
3. L. Hagne, D. Puzosovic, A. Dragoneas and M. Grell, *Sc. Adv. Mater.*, **3**, 907 (2011)
4. A. Dragoneas, M. Grell, M. Hampton, J. E. Macdonald, *Sensor Letters*, (submitted - 2012)
5. L. A. Majewski, R. Schroeder, M. Grell, *J. Phys. D: Appl. Phys.*, **37**, 21-24 (2004)
6. J. Huang, S. Virji, B. H. Weller, R. B. Kaner, *J.-R.C.S.*, **125**, 314 (2003)
7. S. Hoshino *et al.*, *J. Appl. Phys.*, **95**, 5088 (2004)

We would like to thank and acknowledge the contribution of the European Union for the research fellowships of Antonis Dragoneas under the 'FlexSMELL' Marie Curie Initial Training Network of the Seventh Framework Programme. Information: <http://www.flexsmell.eu>